Term Test-1

Answer all questions in the Exam Booklets. Put your name and student number on all exam booklets. You may use a non-programmable calculator. Draw <u>structures</u> and <u>diagrams</u> where appropriate.

The total number of marks is 61 and you have 75 minutes to complete the exam.

Answer all questions.

1. (8) Draw the chemical structure at pH 7 of the following oxidation with HOCl.

- 2. (4) Explain the difference between normal phase and reverse-phase chromatography.
- 3. (8) Describe affinity tag purification of a protein using glutathione-S transferase.
- 4. (15) Outline a protocol for amino acid analysis of a protein and describe in chemical detail peptide hydrolysis by strong acid. What problem arises in amino acid analysis of proteins containing β-branched dipeptides? What can be done about this?
- 5. (6) A 35 micromolar solution of a newly discovered protein, Euphorin, has an absorption at 280 nm of 1.40 and an absorption at 288 nm of 0.96, in a cell of 1 cm path length. The extinction coefficients (in M⁻¹· cm⁻¹) for tyrosine and tryptophan at the two wavelengths are listed below. From the information given, calculate the number of Tyrosine and Tryptophan residues in Euphorin. For full marks show your work.

	$\mathbf{\epsilon}_{280}$	$\boldsymbol{\epsilon}_{288}$
Trp	5690	4815
Tyr	1280	385

6. **(6)** Show the mechanism by which oxidized glutathione can assist in the formation of a disulphide bond in a protein. What is the name of this reaction? What can suppress it?

- 7. (4) Describe how a magnetic sector **OR** a time-of-flight mass spectrometer works.
- 8. **(6)** Describe Fast-Atom Bombardment **OR** Matrix-Assisted Laser Desorption Ionization, **OR** Electrospray Ionization.
- 9. **(4)** Why is the monoisotopic peak in the mass spectrum of a small peptide the most abundant peak whereas the monoisotopic peak is only a minor peak in the mass spectrum of a large protein?

Bonus Question:

10. (2) What is the deadliest known protein and what is its approximate LD_{50} ?