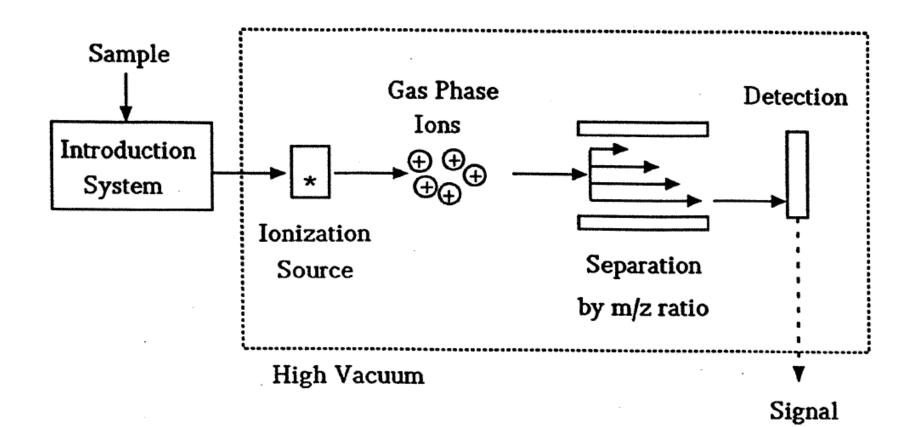
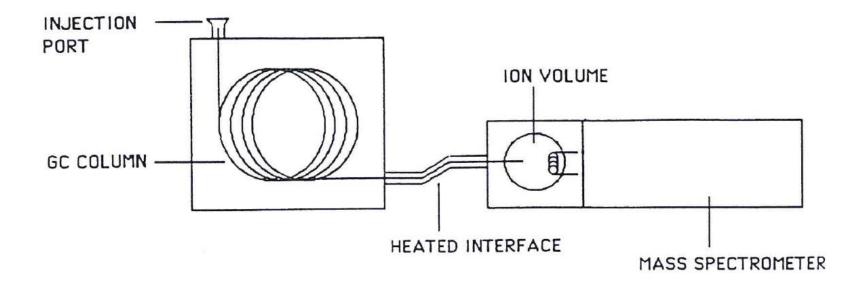


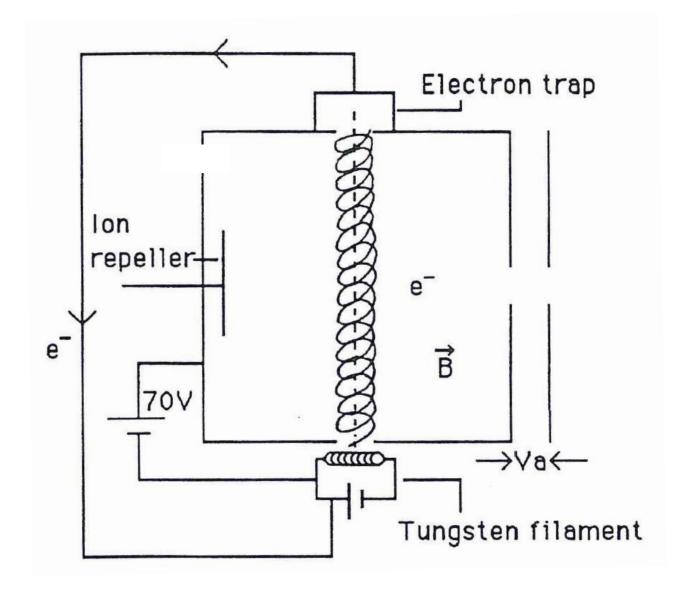
Mass spectrometry



Components of a GC/MS system



The electron impact ionization source

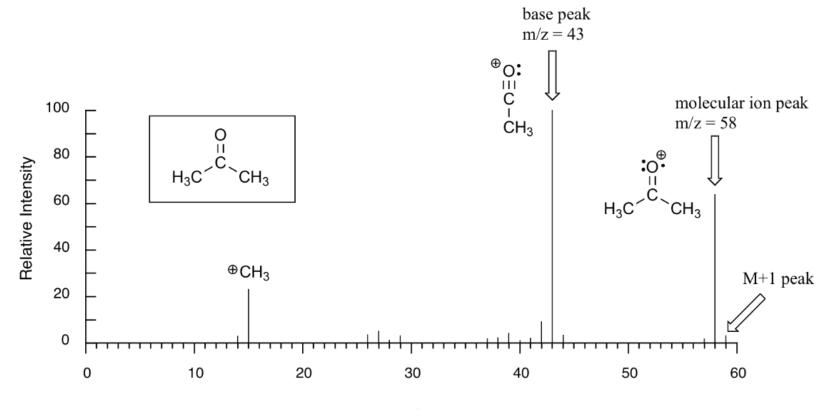


```
M + e^{-} \rightarrow M^{+} + 2e^{-}
```

 $M^{+} \rightarrow F^{+} + N^{-}$

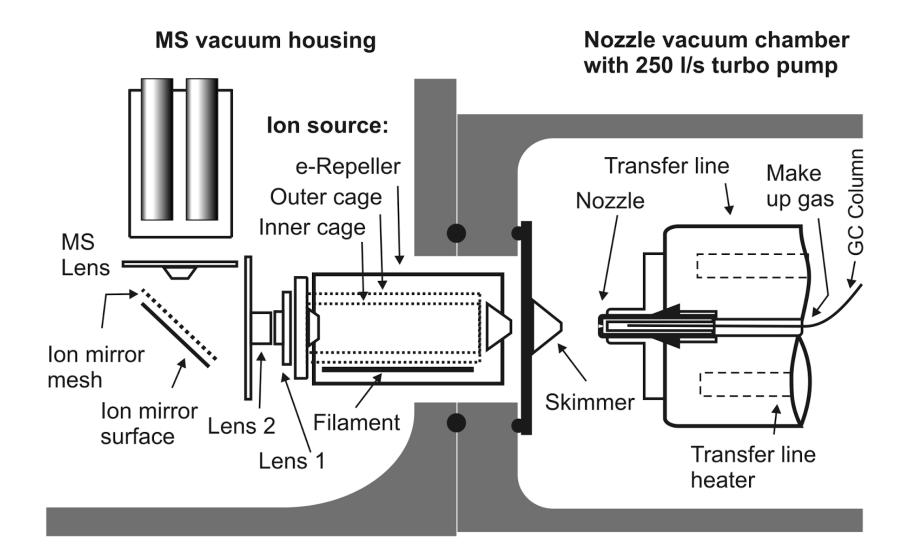
-Sample evaporated in source from heated probe -Sample introduced by GC -Gas inlet

Typical El mass spectrum: acetone



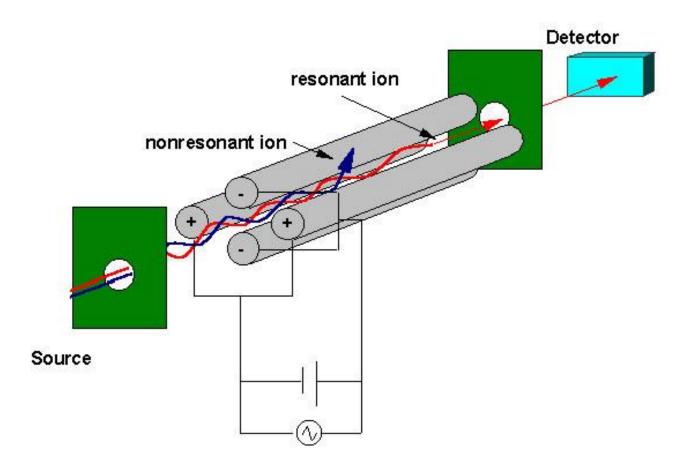
m/z

GC/MS interface



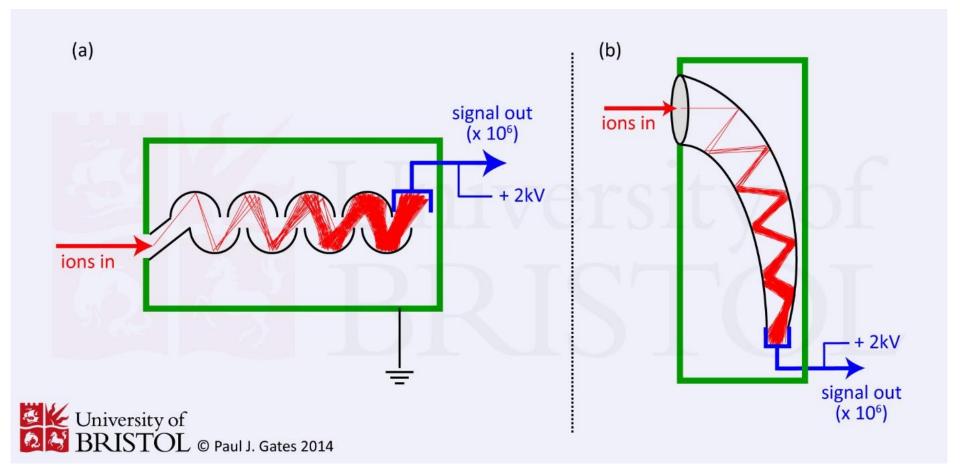
http://blog.avivanalytical.com/2013/06/what-can-be-improved-in-gc-ms-when.html

Quadrupole mass analyzer

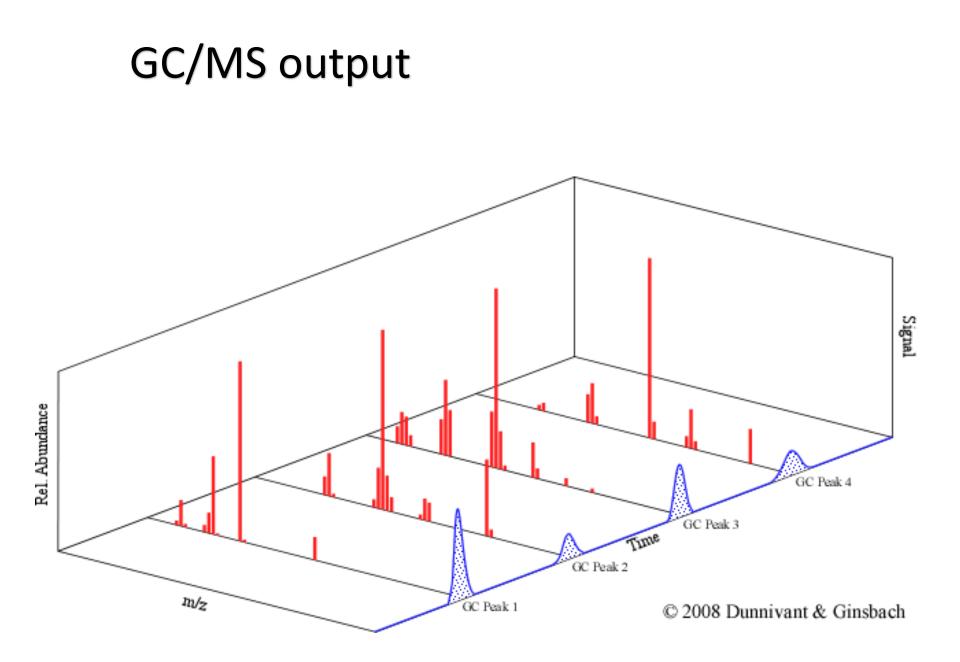


http://www.ivv.fraunhofer.de/en/leistungsangebot/stoerstoffanalyse/ms_pages/Introduction_into_mass_spectrometry/ms_intro_analyzers.html

Detector: electron multiplier



http://www.chm.bris.ac.uk/ms/detectors.xhtml



Application 1

The Preparation and Analysis of Polycyclic Aromatic Hydrocarbons in Meat by GC/MS

APPLICATION NOTE

Gas Chromatography/ Mass Spectrometry

Authors

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Beijing Municipal Center for Food Safety Beijing, China

http://www.perkinelmer.com/PDFs/downloads/APP_PAHinMeatbyGCMS.pdf



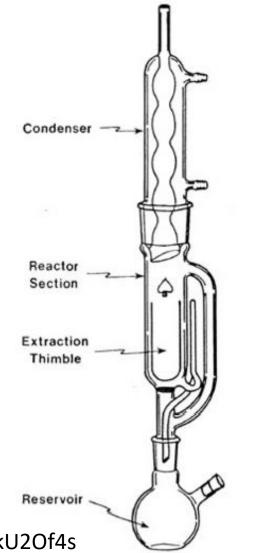
- This application note presents a method developed to measure 15 + 1 EU priority PAHs at low levels using gas chromatography mass spectrometry (GC/MS).
- It also describes a reliable procedure for extraction and purification of PAHs from meat samples.
- The sample preparation focuses on benzo[a]pyrene.
- In addition to method optimization and calibration, a variety of meat samples are analyzed and the amount of PAHs determined.

Experimental

1) Solvent Extraction

A solvent extractor was used to obtain PAHs from the meat samples.

Typically: Soxhlet extraction



https://www.youtube.com/watch?v=tUDdkU2Of4s

2) Gel permeation chromatography (GPC)

• Molecules are separated by size

•Small molecules enter pores of stationary phase (b), large molecules do not (a)

• Larger molecules elute first

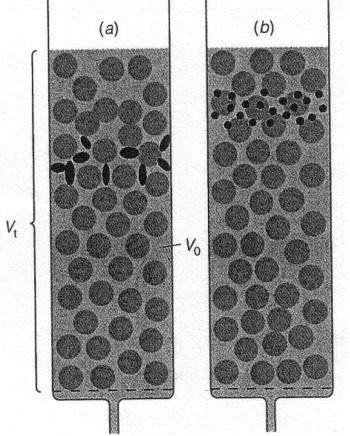


Table 2. Instrumental Parameters.					
Gas Chromatograph PerkinElmer Clarus 680 GC					
Oven Program Initial Temperature	50 °C				
Hold Time 1	2 min				
Ramp 1	25 °C/min to 200 °C				
Hold Time 2	0 min				
Ramp 2	15 °C/min to 310 °C				
Hold Time 3	19.67 min				
Equilibration Time	0.2 min				
Column	PerkinElmer – Elite™-5ms 30 m x 0.25 mm x 0.25 μm				
Injector	Programmable Split/Splitless				
Injection Mode	Splitless and Pressure Pulse				
Injection Volume	1 µL				
Inlet Temperature	280 °C				
Carrier Gas	Helium				
Carrier Gas Flow Rate	1 mL/min				
Mass Spectrometer	PerkinElmer Clarus 600 MS				
Mass Range	45-450 u				
Solvent Delay Time	6 min				
Scan Time	0.20 sec				
Transfer Line Temperature	280 °C				
Source Temperature	240 °C				
SIM Mode	8 SIM groups				
SIR Dwell Time	0.04 sec				

Following the calibration of the system, 5 g of bacon, preserved ham and sausage were analyzed and the PAH concentrations quantified.

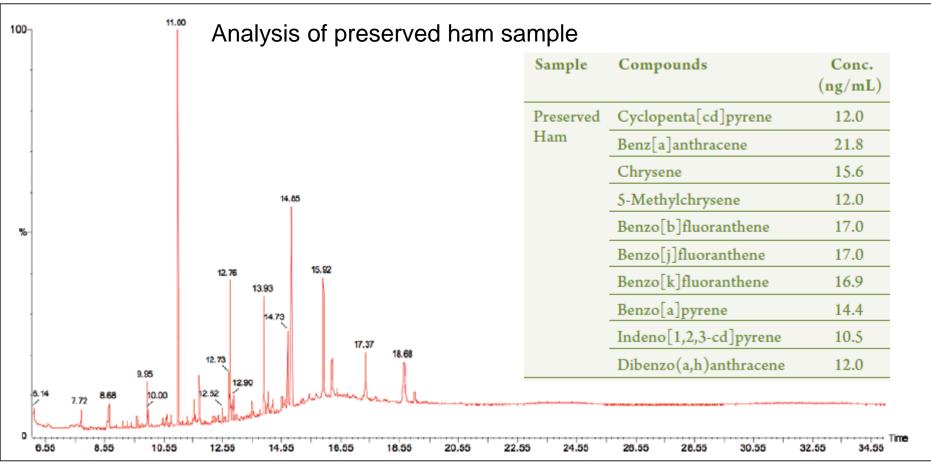
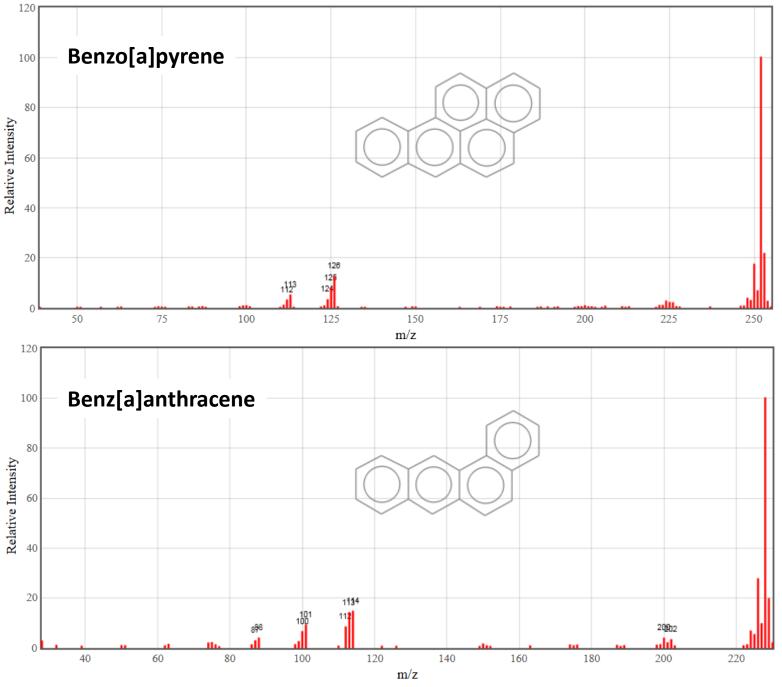


Figure 2. Total ion chromatogram (TIC) of the analysis of preserved ham sample.

Sample	Compounds	Conc. (ng/mL)			
Bacon	Cyclopenta[cd]pyrene	9.3	Sausage	Cyclopenta[cd]pyrene	7.8
	Benz[a]anthracene	22.1		Benz[a]anthracene	21.6
	Chrysene	18.4		Chrysene	11.3
	5-Methylchrysene	9.5		5-Methylchrysene	6.3
	Benzo[b]fluoranthene	15.2		Benzo[b]fluoranthene	12.8
	Benzo[j]fluoranthene	15.2		Benzo[j]fluoranthene	12.8
	Benzo[k]fluoranthene	15.1		Benzo[k]fluoranthene	12.7
	Benzo[a]pyrene	14.2		Benzo[a]pyrene	12.0
	Indeno[1,2,3-cd]pyrene	11.7		Indeno[1,2,3-cd]pyrene	11.6
	Dibenzo(a,h)anthracene	14.3		Dibenzo(a,h)anthracene	14.4

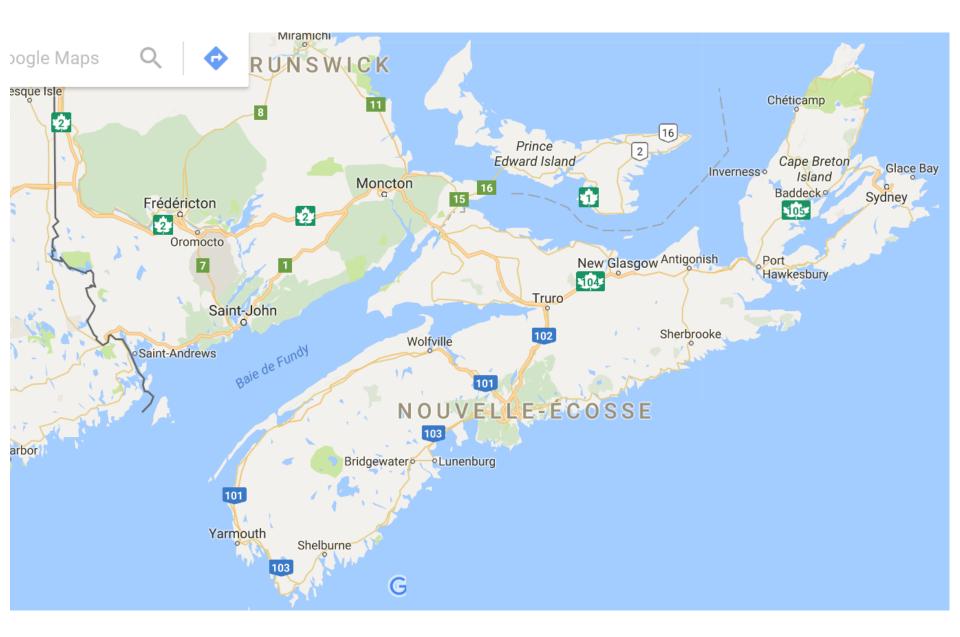


http://webbook.nist.gov/cgi/cbook.cgi?ID=C50328&Mask=200

Name	Retention Time	Quantifier Ion	Qualifier Ion 1	Qualifier Ion 2	%RSD of Peak Area (n=7) 100 ng/mL	r ² (5 – 500 ppb)
Naphthalene-D8	7.28	136	137	108		
Phenanthrene-D10	10.79	188	189	80		
Cyclopenta[cd]pyrene	14.78	226	227	224	0.5	0.9999
Benz[a]anthracene	14.78	228	229	226	1.1	0.9996
Chrysene-D12	14.79	240	241	236		
Chrysene	14.83	228	229	226	3.0	0.9999
5-Methylchrysene	15.52	242	241	239	1.5	0.9997
Benzo[b]fluoranthene	16.56	252	253	250	0.3	0.9986
Benzo[j]fluoranthene	16.56	252	253	250	0.3	0.9986
Benzo[k]fluoranthene	16.56	252	253	250	0.3	0.9987
Benzo[a]pyrene	17.19	252	253	250	0.3	0.9981
Perylene-D12	17.29	264	265	260		
Indeno[1,2,3-cd]pyrene	19.89	276	277	138	0.4	0.9972
Dibenzo(a,h)anthracene	19.96	278	279	276	0.1	0.9965
Benzo[ghi]perylene	20.65	276	277	138	0.9	0.9997
Dibenzo[a,l]pyrene	25.09	302	151	150	0.3	0.9959
Dibenzo[a,e]pyrene	26.85	302	151	150	0.2	0.9942
Dibenzo[a,i]pyrene	27.56	302	151	150	0.2	0.9913
Dibenzo[a,h]pyrene	27.94	302	151	150	0.2	0.9926

Application 2

The Nova Scotia tar pond disaster

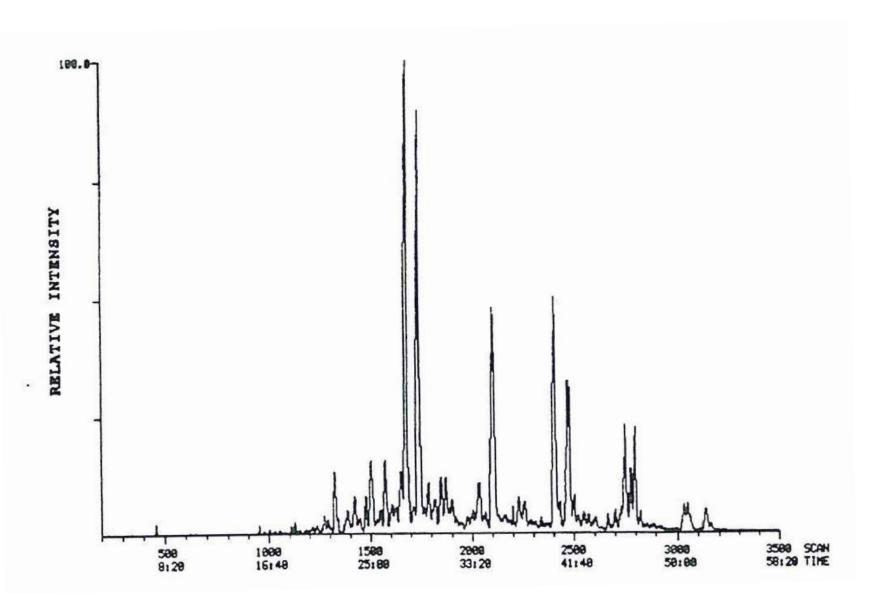


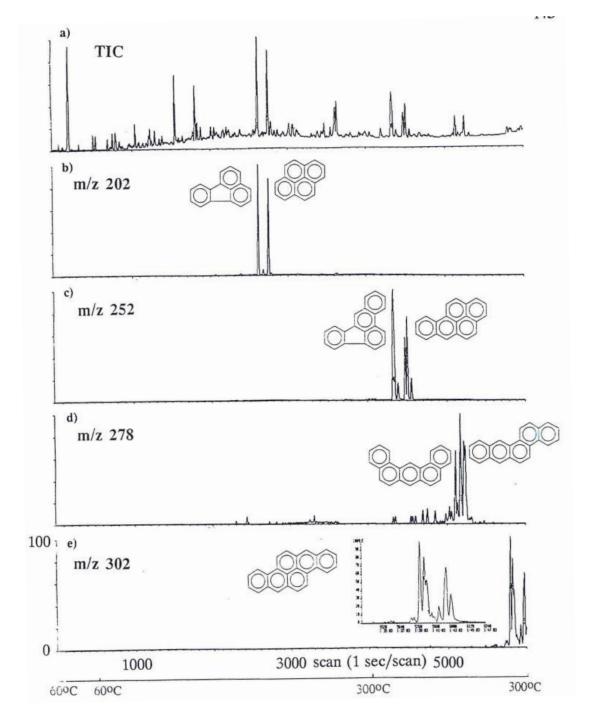




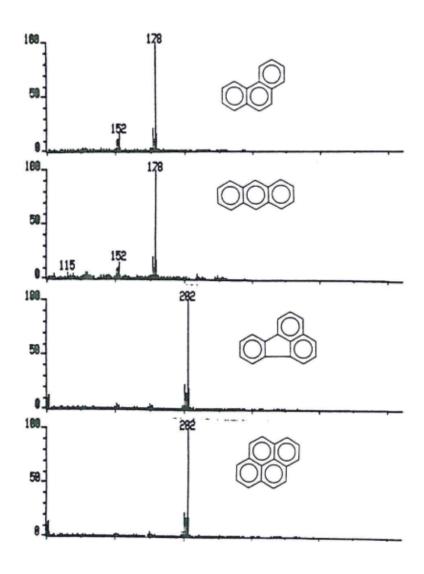


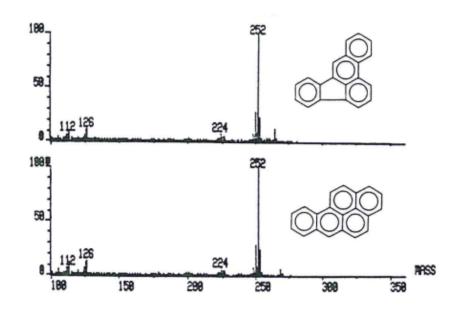
Sediment PAC fraction GC/MS TIC



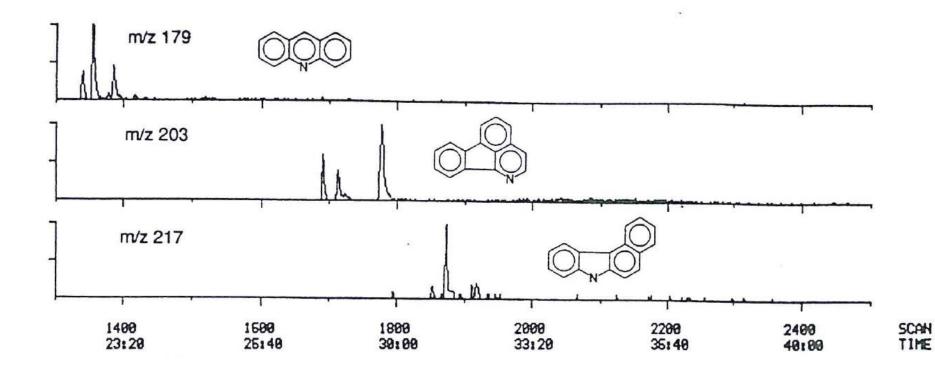








Selected ion chromatograms: PANHs



Remediated Sydney tar ponds unveiled as green space Open Hearth Park is a 39-hectare green area featuring sports fields, walking trails, and a playground

The Canadian Press Posted: Aug 30, 2013 7:04 AM AT Last Updated: Aug 30, 2013 3:26 PM AT



Ways of targeting specific analytes in GC/MS:

- Selected ion chromatograms for M^{,+} or F⁺
- Selected reaction monitoring $(\Delta m/z \text{ when } M^{,+} \rightarrow F^{+})$
- Constant neutral loss



- One of the most widely used analytical technique in all fields
- Combines the advantages of universal (TIC) and specific (SICs) detection
- Qualitative and quantitative
- Relatively low cost in the world of mass spectrometry
- Quadrupole and ion trap mass analyzers mostly used
- Used for volatile compounds and derivatization required for others
- Up to mass about 800-1000.



1: What is the difference between the total ion current and a selected ion current in gas chromatographymass spectrometry (GC/MS)?

2:

a) Explain, in a reaction, how ions are formed by electron impact ionization.

b) What types of molecules is this technique used for?

3: How does an electron capture detector create a signal when analyzing halogenated compounds by GC?