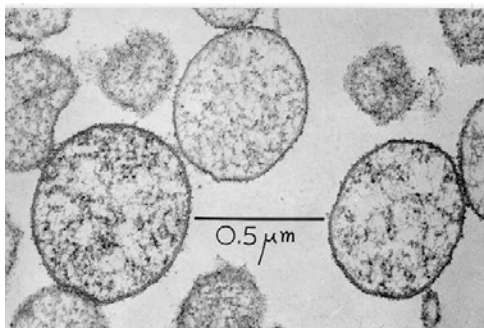


Chapter 9 - Biological Membranes



www.gsbs.utmb.edu/microbook/ch037.htm Mycoplasma

Membranes form a semi-permeable boundary between a cell and its environment.

Membranes also permit subcellular organization and segregation of metabolic pathways.

Membrane-bound Organelles

Mitochondrion

Endoplasmic Reticulum

Nucleus

Peroxisome

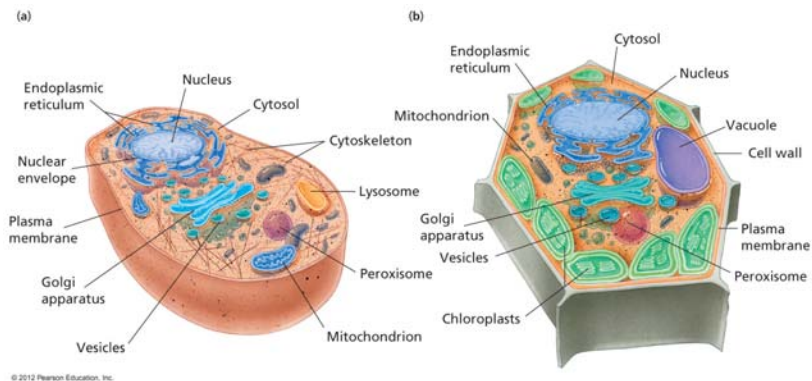
Vacuole

Lysosome

Golgi Apparatus

Secretory Vesicle

Chloroplast



Composition: Lipids, Proteins, Sterols, Carbohydrates.

Myelin Sheath is mostly lipid. Mitochondrial membrane is mostly protein so there is lots of variety in composition.

Lipid composition depends on Kingdom, species, organelle, leaflet.

	Outer	Inner
P-Choline	22%	9%
P-Ethanolamine	7%	24%
P-Serine	0%	9%
Sphingomyelin	20%	7%

Carbohydrates are abundant on plasma membrane but rare on intracellular membranes.

Fluid Mosaic Model

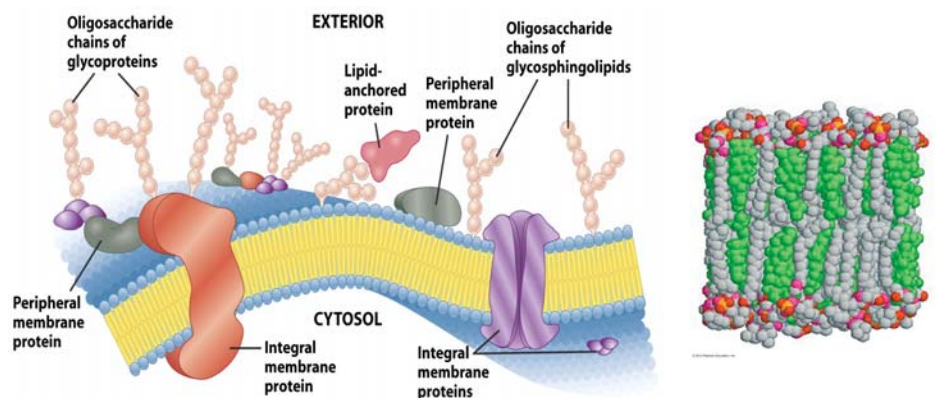


Figure 9-21 Principles of Biochemistry, 4/e
© 2006 Pearson Prentice Hall, Inc.

A lipid bilayer, 5-8 nm thick.

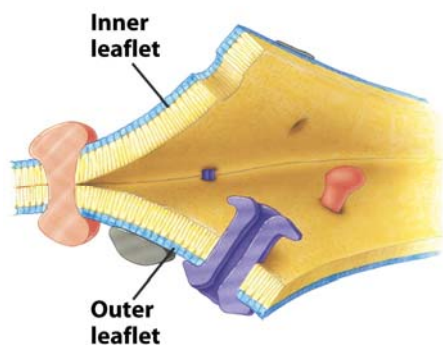


Figure 9-24a Principles of Biochemistry, 4/e
© 2006 Pearson Prentice Hall, Inc.

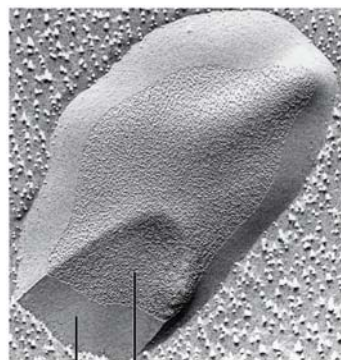


Figure 9-24b Principles of Biochemistry, 4/e
© 2006 Pearson Prentice Hall, Inc.

Lipid can diffuse laterally – in a red blood cell circumnavigation takes a few seconds.

Transbilayer diffusion (flip-flop) is rare.

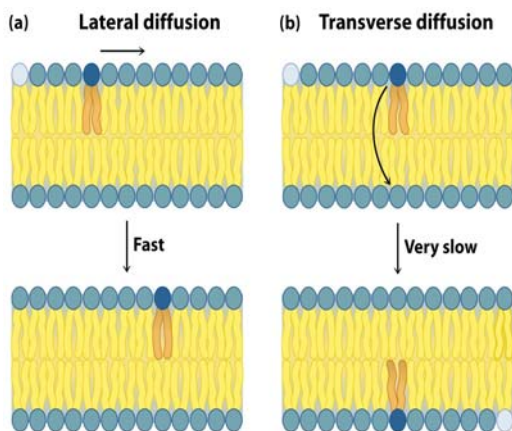


Figure 9-22 Principles of Biochemistry, 4/e
© 2006 Pearson Prentice Hall, Inc.

Special enzymes, **flipases**, exist to assist this process.

Bilayers undergo a phase transition:

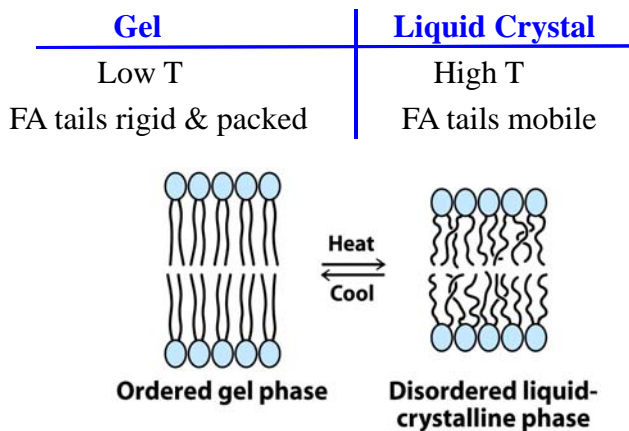


Figure 9-23 Principles of Biochemistry, 4/e
© 2004 Pearson Prentice Hall, Inc.

In living cells the phase transition is broad and may be between 10° and 40°C.

The fluidity of a membrane depends on the amount of saturated and unsaturated lipid, and the amount of sterol present.

Membrane Proteins:

1. Peripheral Membrane Proteins associate with the surface of the bilayer.

They are held in place by weak interactions (H-bonds, van der Waal's, electrostatic) or covalent attachment to the lipid.

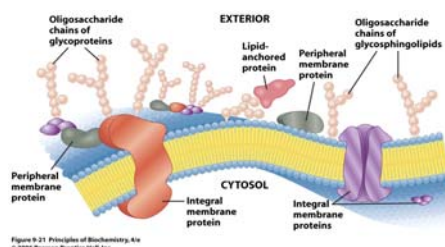


Figure 9-21 Principles of Biochemistry, 4/e
© 2004 Pearson Prentice Hall, Inc.

High salt concentrations may release them from membrane.

An example is Cytochrome c.

2. **Integral Membrane Proteins** span the bilayer.

Their external surfaces are **hydrophilic**.

They are in contact with water in the cytoplasm or outside the cell.

Their internal surfaces are **hydrophobic**, in contact with lipid.

Removal from the membrane is usually by detergent extraction.

Classes Of Integral Membrane Proteins

a. Traverse membrane as a single α -helix.

b. Traverse membrane as 4-16 α -helices.

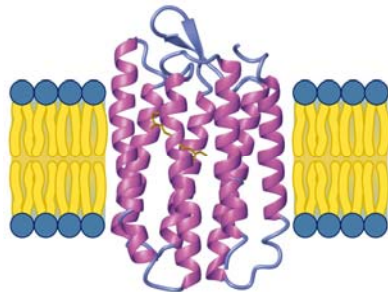
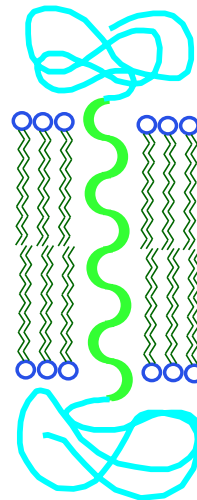


Figure 9-27 Principles of Biochemistry, 4th
© 2004 Pearson Prentice Hall, Inc.



- c. Traverse membrane as 8-16 β -strands forming a giant β -barrel – **Porins**

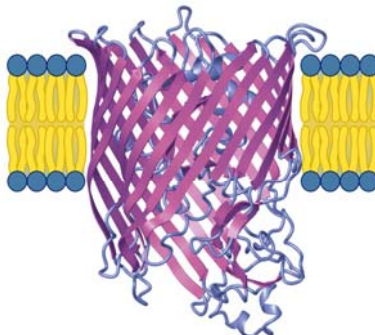


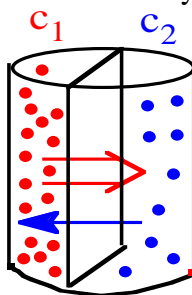
Figure 9-28 Principles of Biochemistry, 4/e
© 2004 Pearson Education, Inc.

Some membrane proteins can diffuse laterally in the plane of the bilayer – others are anchored to the cytoskeleton.

Membranes are tough, but flexible. They can change shape in growth and movement, and are able to seal temporary leaks.

Transport Across Bilayers

The 2nd law of thermodynamics states that systems tend to the greatest entropy, disorder.



Diffusion across a semipermeable membrane occurs in both directions, but net diffusion is from **1** \rightarrow **2** until $c_1 = c_2$.

CH_4 , O_2 , N_2 , H_2O , NO , CO cross membranes freely, equalizing their concentrations on both sides of the membranes.

NO and CO are second messengers.

This is called "Simple Diffusion" and is governed by thermodynamics.

When solute concentrations are different on 2 sides of a membrane there must also be a H₂O concentration gradient.

H₂O crosses the membrane until the osmotic strength is equal on both sides.

Some cells, *e.g.* kidney and lung, use integral membrane proteins called aquaporins to increase the rapid transport of water across membranes. They form a pore that permits rapid diffusion of the water down its concentration gradient. This is called:

Facilitated Diffusion:

Many large polar molecules, including carbohydrates, and amino acids are transported this way.

The transport proteins are called permeases.

e.g. Red Blood Cell Glucose Permease

$M_r \sim 45,000$ Da 12 α -helices form a pore that transports glucose 50,000 times faster than simple diffusion through the membrane.

Active Transporters can pump solute **against** a concentration gradient.

e.g. The $\text{Na}^+ - \text{K}^+$ ATPase uses energy from the hydrolysis of ATP to pump 2 K^+ into a cell and 3 Na^+ out of the cell.

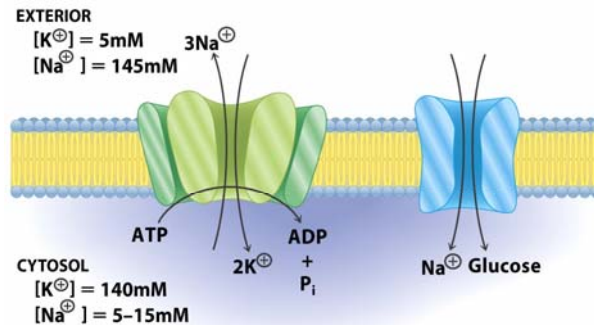


Figure 9-35 Principles of Biochemistry, 4/e
 © 2006 Pearson Prentice Hall, Inc.

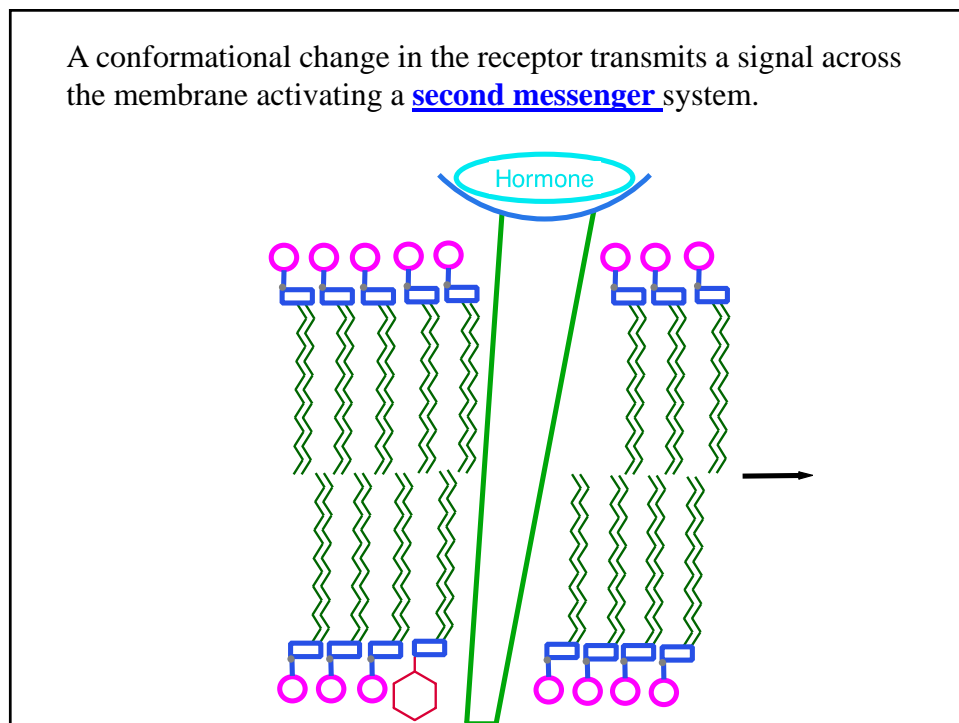
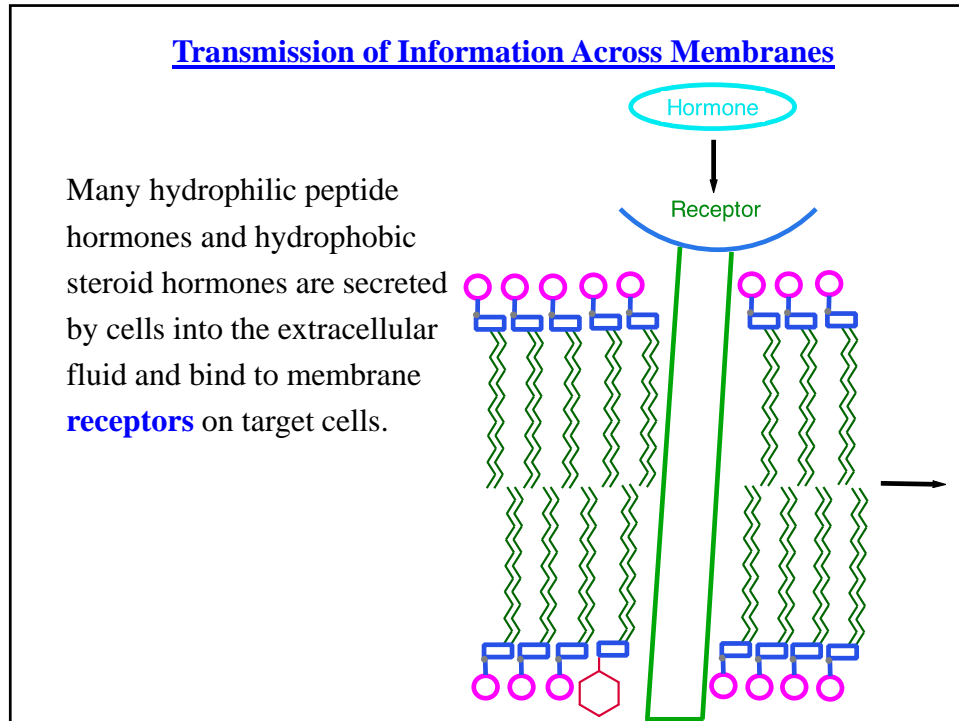
For most animal cells: $[\text{Na}^+]_{\text{in}} < [\text{Na}^+]_{\text{out}}$ and $[\text{K}^+]_{\text{in}} > [\text{K}^+]_{\text{out}}$

There is a net movement of positive charge out of the cell, which then becomes negatively charged.

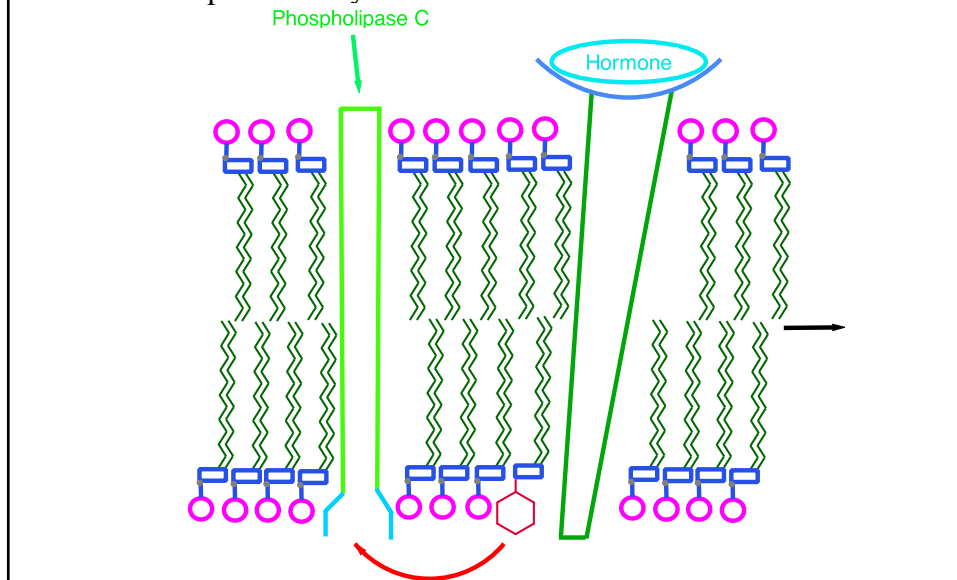
The transmembrane electrical potential is:

$$\Delta\Psi = 50 - 70 \text{ milliVolts}$$

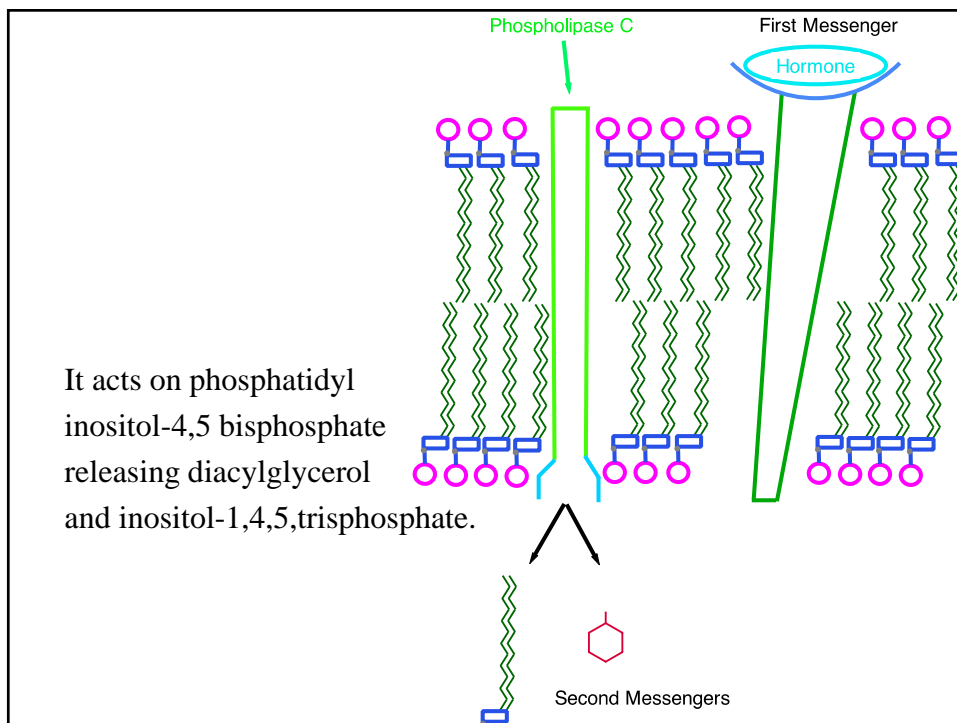
Up to 25% of the resting metabolic energy of a cell is used to maintain this gradient.



For example, the hormone epinephrine (adrenaline) binds to adrenergic receptors activating phospholipase C an integral membrane protein enzyme.



It acts on phosphatidyl inositol-4,5 bisphosphate releasing diacylglycerol and inositol-1,4,5, trisphosphate.



Both are intracellular **Second Messengers** that regulate enzyme activity and metabolism in **cell signalling** aka **signal transduction**.

Another example: Insulin binds to the insulin receptor and activates an enzyme cascade that leads to changes in sugar and lipid metabolism.